

Quinine Sulfate Capsules

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Expert Committee Chemical Medicines Monographs 1

Reason for Revision Compliance

In accordance with the Rules and Procedures of the 2015–2020 Council of Experts, the Chemical Medicines Monographs 1 Expert Committee has revised the Quinine Sulfate Capsules monograph. The purpose for the revision is to add *Dissolution Test 3* to accommodate FDA-approved drug products with different dissolution conditions than the existing dissolution tests.

The Quinine Sulfate Capsules Revision Bulletin supersedes the currently official monograph.

Quinine Sulfate Capsules

DEFINITION

Quinine Sulfate Capsules contain amounts of quinine sulfate and dihydroquinine sulfate totaling NLT 90.0% and NMT 110.0% of the labeled amount of quinine sulfate, calculated as $(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O$.

IDENTIFICATION

• A.

Sample: Nominally 100 mg of quinine sulfate from the contents of Capsules

Analysis: Shake the Sample with 100 mL of dilute sulfuric acid (1 in 350), and filter.

Acceptance criteria: An appropriate dilution of the filtrate exhibits a vivid blue fluorescence. On the addition of a few drops of hydrochloric acid, the fluorescence disappears.

• **B.** The R_F value of the principal spot from the *Sample solution* corresponds to that from *Standard solution* A, as obtained in the test for *Organic Impurities*.

• C. <u>Identification Tests—General (191), Chemical Identification Tests, Sulfate</u>

Sample: Nominally 20 mg of quinine sulfate from the contents of Capsules

Analysis: Shake the Sample with 10 mL of dilute hydrochloric acid (1 in 100), and filter.

Acceptance criteria: The filtrate meets the requirements.

• **D.** The retention time of the major peak of the *Sample solution* corresponds to that of the *Standard solution*, as obtained in the *Assay*.

ASSAY

• PROCEDURE

Solution A: Add 35.0 mL of <u>methanesulfonic acid</u> to 20.0 mL of glacial acetic acid, and dilute with <u>water</u> to 500 mL.

Solution B: Dissolve 10.0 mL of <u>diethylamine</u> in <u>water</u> to obtain 100 mL of solution.

Mobile phase: Acetonitrile, Solution A, Solution B, and water (100:20:20:860). Adjust with Solution B to a pH of 2.6 if the pH is found to be lower.

System suitability solution: 0.2 mg/mL each of <u>USP Quinine Sulfate RS</u> and dihydroquinine, dissolved in 10% of the final volume of methanol. Dilute with *Mobile phase* to volume.

Standard solution: 0.2 mg/mL of USP Quinine Sulfate RS in Mobile phase

Sample stock solution: Nominally 1.6 mg/mL of quinine sulfate in methanol prepared as follows. Transfer an amount, equivalent to 160 mg of quinine sulfate from the contents of NLT 20 Capsules, to a 100-mL volumetric flask, add 80 mL of <u>methanol</u>, and shake the flask by mechanical means for 30 min. Dilute with <u>methanol</u> to volume, and filter, discarding the first 10 mL of the filtrate.

Sample solution: Nominally 0.2 mg/mL of quinine sulfate in *Mobile phase* from the *Sample stock solution* **Chromatographic system**

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 235 nm

Column: 3.9-mm × 30-cm; packing L1

Flow rate: 1 mL/min Injection volume: 50 µL

System suitability

Sample: System suitability solution

[Note—The relative retention times for quinine and dihydroquinine are 1 and 1.5, respectively.]

Suitability requirements

Resolution: NLT 1.2 between quinine and dihydroquinine

Relative standard deviation: NMT 2.0% for the quinine peak

Analysis

Samples: Standard solution and Sample solution

Calculate the percentage of the labeled amount of quinine sulfate $[(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O]$ calculated as the sum of quinine sulfate and dihydroquinine sulfate in the portion of Capsules taken:

Result =
$$[(r_{b,U} + r_{d,U})/(r_{b,S} + r_{d,S})] \times (C_S/C_U) \times 100$$

 $r_{b,U}$ = peak area response of quinine from the Sample solution

 $r_{d.U}$ = peak area response of dihydroquinine from the Sample solution

 $r_{b.S}$ = peak area response of quinine from the *Standard solution*

 $r_{d,S}$ = peak area response of dihydroquinine from the *Standard solution*

 C_S = concentration of <u>USP Quinine Sulfate RS</u> in the *Standard solution* (mg/mL)

 C_{II} = nominal concentration of quinine sulfate in the Sample solution (mg/mL)

Acceptance criteria: 90.0%-110.0%

PERFORMANCE TESTS

Change to read:

● **Dissolution** (711)

Test 1

Medium: 0.1 N hydrochloric acid; 900 mL

Apparatus 1: 100 rpm

Time: 45 min

Detection: UV maximum at about 248 nm

Standard solution: Prepare a solution of known concentration of <u>USP Quinine Sulfate RS</u> in *Medium*.

Sample solution: A filtered portion of the solution under test, suitably diluted with Medium

Analysis

Samples: Standard solution and Sample solution

Calculate the percentage of the labeled amount of quinine sulfate $[(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O]$ dissolved.

Tolerances: NLT 75% (Q) of the labeled amount of quinine sulfate $[(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O]$ is dissolved.

Test 2: If the product complies with this test, the labeling indicates that the product meets USP *Dissolution Test 2*.

Medium: 0.1 N hydrochloric acid; 900 mL

Apparatus 1: 100 rpm

Time: 30 min

Solution A: Add 7.0 mL of <u>methanesulfonic acid</u> to 4.0 mL of <u>glacial acetic acid</u>, and dilute with <u>water</u> to 100 mL.

Solution B: Dissolve 10.0 mL of <u>diethylamine</u> in <u>water</u> to obtain 100 mL of solution.

Mobile phase: Water, acetonitrile, Solution A, and Solution B (81:15:2:2). Adjust with Solution B to a pH of 3.0.

Standard solution: Prepare a solution of known concentration of <u>USP Quinine Sulfate RS</u> in *Medium*.

Sample solution: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size, and suitably dilute with *Medium*.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 235 nm

Column: 4.6-mm \times 15-cm; 5- μ m packing <u>L1</u>

Flow rate: 1.2 mL/min Injection volume: 10 μ L

System suitability

Sample: Standard solution **Suitability requirements**

Tailing factor: NMT 2.0 for the quinine peak

Relative standard deviation: NMT 2.0% for the sum of quinine and dihydroquinine peaks

Analysis

Samples: Standard solution and Sample solution

Calculate the percentage of the labeled amount of quinine sulfate $[(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O]$ dissolved.

Result = $(r_1/r_S) \times (C_S/L) \times (M_{r_1}/M_{r_2}) \times D \times V \times 100$

 r_{II} = sum of the peak responses of quinine and dihydroquinine from the Sample solution

 r_S = sum of the peak responses of quinine and dihydroquinine from the Standard solution

 C_S = concentration of <u>USP Quinine Sulfate RS</u> in the *Standard solution* (mg/mL)

L = label claim (mg/Capsule)

 M_{r1} = molecular weight of quinine sulfate, 782.94

 M_{r2} = molecular weight of anhydrous quinine sulfate, 746.92

D = dilution factor of the Sample solution

V = volume of Medium, 900 mL

Tolerances: NLT 80% (Q) of the labeled amount of quinine sulfate $[(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O]$ is dissolved.

Test 3: If the product complies with this test, the labeling indicates that the product meets USP Dissolution

Test 3.

Medium: 0.1 N hydrochloric acid; 900 mL

Apparatus 2: 50 rpm

Time: 30 min

Standard solution: 0.0144 mg/mL of USP Quinine Sulfate RS in Medium

Sample solution: A filtered portion of the solution under test, suitably diluted with Medium

Instrumental conditions

Mode: UV

Analytical wavelength: 248 nm

Cell: 1 cm
Blank: Medium

Analysis

Samples: Standard solution and Sample solution

Calculate the percentage of the labeled amount of quinine sulfate [(C₂₀H₂₄N₂O₂)₂·H₂SO₄·2H₂O]

dissolved:

Result =
$$(r_U/r_S) \times (C_S/L) \times (M_{r_1}/M_{r_2}) \times D \times V \times 100$$

 r_{II} = absorbance of the Sample solution

 $r_{\rm S}$ = absorbance of the Standard solution

 C_S = concentration of <u>USP Quinine Sulfate RS</u> in the *Standard solution* (mg/mL)

= label claim (mg/Capsule)

 M_{r1} = molecular weight of quinine sulfate, 782.94

 M_{r2} = molecular weight of anhydrous quinine sulfate, 746.92

D = dilution factor of the Sample solution, if necessary

V = volume of Medium, 900 mL

Tolerances: NLT 75% (Q) of the labeled amount of quinine sulfate [(C₂₀H₂₄N₂O₂)₂·H₂SO₄·2H₂O] is

dissolved. ▲ (RB 1-Aug-2020)

• Uniformity of Dosage Units (905)

Procedure for content uniformity

Diluent: Hydrochloric acid (1 in 100)

Standard solution: 40 µg/mL of USP Quinine Sulfate RS in Diluent

Sample solution: Transfer the contents of one Capsule to a 250-mL volumetric flask, add 175 mL of *Diluent*, and shake by mechanical means for 30 min. Add *Diluent* to volume. Filter a portion of the mixture, discarding the first 20 mL of the filtrate.

Instrumental conditions

(See <u>Ultraviolet-Visible Spectroscopy (857)</u>.)

Mode: UV Cell: 1 cm

Analytical wavelength: Maximum at about 345 nm

Blank: Water
Analysis

Samples: Standard solution and Sample solution

Calculate the percentage of the labeled amount of quinine sulfate $[(C_{20}H_{24}N_2O_2)_2 \cdot H_2SO_4 \cdot 2H_2O]$ in the Capsule taken:

Result =
$$(A_U/A_S) \times (C_S/C_U) \times 100$$

 A_{II} = absorbance of the Sample solution

 A_S = absorbance of the *Standard solution*

 C_S = concentration of <u>USP Quinine Sulfate RS</u> in the *Standard solution* (mg/mL)

 C_{II} = nominal concentration of quinine sulfate in the Sample solution (mg/mL)

Acceptance criteria: Meet the requirements

IMPURITIES

• ORGANIC IMPURITIES

Standard stock solution: 6 mg/mL of USP Quinine Sulfate RS in diluted alcohol

Standard solution A: 0.06 mg/mL of <u>USP Quinine Sulfate RS</u> from the *Standard stock solution* in diluted alcohol

Standard solution B: 0.05 mg/mL of <u>USP Quininone RS</u> (corresponding to 0.06 mg/mL of the sulfate) and 0.10 mg/mL of cinchonidine (corresponding to 0.12 mg/mL of the sulfate) in diluted alcohol

Sample solution: Nominally 6 mg/mL of quinine sulfate in diluted alcohol prepared as follows. Shake the equivalent of 150 mg of quinine sulfate from the contents of Capsules with 25 mL of diluted alcohol for 10 min, and filter.

Chromatographic system

(See <u>Chromatography (621), General Procedures, Thin-Layer Chromatography</u>.)

Mode: TLC

Adsorbent: 0.25-mm layer of chromatographic silica gel mixture

Application volume: 10 µL

Developing solvent system: Chloroform, acetone, and diethylamine (50:40:10). [Note—The solvent

chamber being used without previous equilibration.]

Analysis

Samples: Standard solution A, Standard solution B, and Sample solution

Proceed as directed in <u>Chromatography (621)</u>, <u>General Procedures</u>, <u>Thin-Layer Chromatography</u>. Allow the spots to dry, and develop the chromatogram using a solvent chamber without previous equilibration. When the solvent front has moved about 15 cm, remove the plate from the chamber, mark the solvent front, and allow the solvent to evaporate. Locate the spots on the plate by spraying with <u>glacial acetic acid</u>, and examine under long-wavelength UV light.

Acceptance criteria: Any spot produced by the *Sample solution* at the R_F value of a spot produced by *Standard solution B* is not greater in size or intensity than that corresponding spot. Apart from these spots and from the spot appearing at the R_F value of quinine sulfate, any additional fluorescent spot is not greater in size or intensity than the spot from *Standard solution A*. Spray the plate with potassium iodoplatinate TS. Any spot produced by the *Sample solution* is not greater in size or intensity than a corresponding spot from *Standard solution B*.

ADDITIONAL REQUIREMENTS

- PACKAGING AND STORAGE: Preserve in tight containers.
- **LABELING:** When more than one *Dissolution* test is given, the labeling states the *Dissolution* test used only if *Test 1* is not used.
- USP REFERENCE STANDARDS (11)

USP Quinine Sulfate RS
USP Quininone RS
Cinchonan-9-one, 6'-methoxy-, (8α) -. $C_{20}H_{22}N_2O_2$ 322.40

Page Information:

Not Applicable

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