

### Niacin Extended-Release Tablets

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<b>Expert Committee</b>	Non-Botanical Dietary Supplements
<b>Reason for Revision</b>	Compliance

In accordance with the Rules and Procedures of the 2015-2020 Council of Experts, the Non-Botanical Dietary Supplement Expert Committee has revised the Niacin Extended-Release Tablets monograph. The purpose for the revision is to add *Dissolution Test 5* for drug product approved by the FDA.

The Niacin Extended-Release Tablets Revision Bulletin supersedes the currently official monograph. The Revision Bulletin will be incorporated in the *Second Supplement to USP 41-NF 36*.

Should you have any questions, please contact Natalia Davydova (301-816-8328 or [nd@usp.org](mailto:nd@usp.org))

## Niacin Extended-Release Tablets

### DEFINITION

Niacin Extended-Release Tablets contain NLT 90.0% and NMT 110.0% of the labeled amount of niacin ( $C_6H_5NO_2$ ).

### IDENTIFICATION

- **A.** The retention time of the major peak of the *Sample solution* corresponds to that of the *Standard solution*, as obtained in the *Assay*.

### ASSAY

#### PROCEDURE

**Diluent:** Methanol and water (82:18)

**Mobile phase:** Methanol and water (82:18), adjusted with glacial acetic acid to a pH of  $3.15 \pm 0.05$

**Standard solution:** 250  $\mu\text{g/mL}$  of USP Niacin RS, 50  $\mu\text{g/mL}$  of USP 6-Hydroxynicotinic Acid RS, and 97.8  $\mu\text{g/mL}$  of pyridine in *Diluent*

**Sample solution:** Transfer a quantity of powder, equivalent to 50 mg of niacin from NLT 20 finely powdered Tablets, to a suitable flask, add *Diluent*, and stir for 2 h. Dilute with *Diluent* to a final concentration of 250  $\mu\text{g/mL}$  of niacin.

#### Chromatographic system

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC

**Detector:** UV 260 nm

**Column:** 4.6-mm  $\times$  15-cm; 5- $\mu\text{m}$  packing L8

**Flow rate:** 1.0 mL/min

**Injection volume:** 25  $\mu\text{L}$

#### System suitability

**Sample:** *Standard solution*

[NOTE—See *Table 4* for relative retention times.]

#### Suitability requirements

**Resolution:** NLT 1.5 between pyridine and 6-hydroxynicotinic acid, and NLT 1.5 between 6-hydroxynicotinic acid and niacin

**Relative standard deviation:** NMT 3.0% for each of the peaks

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) in the portion of Tablets taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak area of niacin from the *Sample solution*

$r_S$  = peak area of niacin from the *Standard solution*

$C_S$  = concentration of USP Niacin RS in the *Standard solution* (mg/mL)

$C_U$  = nominal concentration of niacin in the *Sample solution* (mg/mL)

**Acceptance criteria:** 90.0%–110.0%

### PERFORMANCE TESTS

#### Change to read:

#### DISSOLUTION (711)

##### Test 1

**Medium:** Water; 900 mL

**Apparatus 1:** 100 rpm

**Times:** 1, 3, 6, 9, 12, and 20 h; without *Medium* replacement. [NOTE—Withdraw the same volume at each time point.]

**Solution A:** Solution of sodium heptanesulfonate in acetic acid, methanol, and water (4:44:33:19), w/w<sup>1</sup>

**Mobile phase:** Mixture of methanol, water, and *Solution A* (560:440:25)

**Standard solution:** USP Niacin RS at a known concentration in water in the range of 75–750  $\mu\text{g/mL}$

**Sample solution:** Filtered portion of the solution under test suitably diluted with *Medium* if necessary

#### Chromatographic system

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC

**Detector:** UV 254 nm

**Column:** 3.9-mm  $\times$  15-cm; 10- $\mu\text{m}$  packing L1

**Flow rate:** 1.0 mL/min

**Injection volume:** 15  $\mu\text{L}$

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0

**Relative standard deviation:** NMT 2.0%

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Determine, in mg/mL, the content of niacin ( $C_6H_5NO_2$ ) in the *Medium* at each time point:

$$\text{Result} = (r_U/r_S) \times C_S \times D$$

$r_U$  = peak area of niacin from the *Sample solution*

$r_S$  = peak area of niacin from the *Standard solution*

$C_S$  = concentration of USP Niacin RS in the *Standard solution* (mg/mL)

$D$  = dilution factor for the *Sample solution*

Calculate the percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at each time point:

At 1 h:

$$\text{Result}_1 = (C_1 \times V/L) \times 100$$

At 3 h:

$$\text{Result}_2 = [C_2 \times (V - V_3) + C_1 \times V_3] \times 100/L$$

At 6 h:

$$\text{Result}_3 = [C_3 \times (V - 2 \times V_3) + (C_1 + C_2) \times V_3] \times 100/L$$

At 9 h:

$$\text{Result}_4 = [C_4 \times (V - 3 \times V_3) + (C_1 + C_2 + C_3) \times V_3] \times 100/L$$

At 12 h:

$$\text{Result}_5 = [C_5 \times (V - 4 \times V_3) + (C_1 + C_2 + C_3 + C_4) \times V_3] \times 100/L$$

At 20 h:

$$\text{Result}_6 = [C_6 \times (V - 5 \times V_3) + (C_1 + C_2 + C_3 + C_4 + C_5) \times V_3] \times 100/L$$

$C$  = as  $C_1, C_2, \dots, C_6$ , the content of niacin in the *Medium* at each time point (mg/mL)

$V$  = volume of *Medium*, 900 mL

$V_3$  = volume of sample withdrawn at each time point (mL)

$L$  = label claim (mg/Tablet)

**Tolerances:** The percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at the times specified in *Table 1*, *Table 2*, and *Table 3* conforms to *Dissolution* (711), *Acceptance Table 2*.

<sup>1</sup> Commercially available from Waters Corporation as PIC B7 Reagent (Part #85103).

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**Table 1. For Tablets Labeled to Contain 500 mg or Less/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 15
3	17–32
6	33–48
9	43–63
12	52–77
20	NLT 75

**Table 2. For Tablets Labeled to Contain 750 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 15
3	16–31
6	31–46
9	42–62
12	51–76
20	NLT 75

**Table 3. For Tablets Labeled to Contain 1000 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 15
3	15–30
6	30–45
9	40–60
12	50–75
20	NLT 75

**Test 2:** If the product complies with this test, the labeling indicates that it meets USP *Dissolution Test 2*.

**Acid stage medium:** 0.1 N hydrochloric acid; 900 mL

**Buffer stage medium:** 6.8 g of monobasic potassium phosphate and 0.89 g of sodium hydroxide pellets in 1000 mL of water. Adjust with diluted sodium hydroxide or phosphoric acid to a pH of 6.8; 900 mL.

**Apparatus 1:** 100 rpm

**Times:** 1, 4, 12, and 24 h: 1 and 4 h in the *Acid stage medium*; 12 and 24 h in the *Buffer stage medium*. Replace the volume withdrawn with the equal volume of medium preheated to  $37 \pm 0.5^\circ$ .

**Procedure:** After 4 h replace the *Acid stage medium* with the *Buffer stage medium*, and run the test for the times specified (additional 20 h for a total of 24 h).

[NOTE—Withdraw the same volume at each time point. Pass a portion of the solution through a suitable filter.]

**Standard stock solution:** 0.2 mg/mL of USP Niacin RS in water

**Standard solution 1:** Dilute *Standard stock solution* with *Acid stage medium* to a final concentration of 0.01 mg/mL of USP Niacin RS.

**Standard solution 2:** Dilute *Standard stock solution* with *Buffer stage medium* to a final concentration of 0.01 mg/mL of USP Niacin RS.

**Sample solution**

**For Tablets labeled to contain 500 mg:** Dilute a filtered portion of the solution under test with appropriate dissolution medium 25-fold.

**For Tablets labeled to contain 750 mg:** Dilute a filtered portion of the solution under test with appropriate dissolution medium 33-fold.

**For Tablets labeled to contain 1000 mg:** Dilute a filtered portion of the solution under test with appropriate dissolution medium 50-fold.

**Instrumental conditions**

(See *Ultraviolet-Visible Spectroscopy* (857).)

**Mode:** UV

**Analytical wavelength:** 262 nm

**Path length:** 1 cm

**Blank:** *Acid stage medium* or *Buffer stage medium*

**Analysis**

**Samples:** *Standard solution 1* or *Standard solution 2* and *Sample solution*

Determine the concentration, in mg/mL, of niacin ( $C_6H_5NO_2$ ) in the sample withdrawn from the vessel at each time point:

$$\text{Result} = [(A_U - A_B)/A_S] \times C_S \times D$$

$A_U$  = absorbance of the *Sample solution*

$A_B$  = absorbance of the *Blank*

$A_S$  = absorbance of the *Standard solution*

$C_S$  = concentration of USP Niacin RS in the *Standard solution* (mg/mL)

$D$  = dilution factor for the *Sample solution*

Calculate the percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at each time point:

At 1 h:

$$\text{Result}_1 = (C_1 \times V) \times 100/L$$

At 4 h:

$$\text{Result}_2 = (C_2 \times V + C_1 \times V_3) \times 100/L$$

At 12 h:

$$\text{Result}_3 = [(C_3 + C_2) \times V + C_1 \times V_3] \times 100/L$$

At 24 h:

$$\text{Result}_4 = [(C_4 + C_2) \times V + (C_1 + C_3) \times V_3] \times 100/L$$

$C$  = as  $C_1, \dots, C_4$ , the content of niacin in the related dissolution medium at each time point (mg/mL)

$V$  = volume of *Medium*, 900 mL

$V_3$  = volume of sample withdrawn at each time point (mL)

$L$  = label claim (mg/Tablet)

**Tolerances:** The percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at the times specified in *Table 4* and *Table 5* conforms to *Dissolution* (711), *Acceptance Table 2*.

**Table 4. For Tablets Labeled to Contain 500 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 25
4	30–50
12	65–85
24	NLT 80

**Table 5. For Tablets Labeled to Contain 750 and 1000 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 25
4	30–50

**Table 5. For Tablets Labeled to Contain 750 and 1000 mg/ Tablet (Continued)**

Time (h)	Amount Dissolved (%)
12	55–75
24	NLT 80

**Test 3:** If the product complies with this test, the labeling indicates that it meets USP *Dissolution Test 3*.

**Medium:** Water; 900 mL

**Apparatus 1:** 100 rpm

**Times:** 1, 3, 6, 9, 12, and 20 h; without *Medium* replacement

[NOTE—Withdraw the same volume at each time point. Pass a portion of the solution through a stainless steel filter.]

**Solution A:** 1 mg/mL of sodium 1-hexanesulfonate monohydrate in water

**Mobile phase:** Mixture of *Solution A*, methanol, and glacial acetic acid (840:150:10)

**Standard solution:** (L/900) mg/mL of USP Niacin RS in *Medium*, where L is the label claim in mg/Tablet

[NOTE—Use sonication for complete dissolution, if necessary.]

**Sample solution:** Filtered portion of the solution under test

**Chromatographic system**

(See *Chromatography* <621>, *System Suitability*.)

**Mode:** LC

**Sample cooler:** 10°

**Detector:** UV 262 nm

**Column:** 3.9-mm × 15-cm; 10-μm packing L1

**Column temperature:** 40°

**Flow rate:** 1.5 mL/min

**Injection volume:** 2 μL

**System suitability**

**Sample:** *Standard solution*

**Suitability requirements**

**Theoretical plates:** NLT 1000

**Tailing factor:** NMT 2.0

**Relative standard deviation:** NMT 2.0%

**Analysis**

**Samples:** *Standard solution* and *Sample solution*  
 Determine the concentration, in mg/mL, of niacin (C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>) in the *Medium* at each time point:

$$\text{Result} = (r_U/r_S) \times C_S$$

$r_U$  = peak area of niacin from the *Sample solution*

$r_S$  = peak area of niacin from the *Standard solution*

$C_S$  = concentration of USP Niacin RS in the *Standard solution* (mg/mL)

Calculate the percentage of the labeled amount of niacin (C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>) dissolved at each time point:

At 1 h:

$$\text{Result}_1 = (C_1 \times V/L) \times 100$$

At 3 h:

$$\text{Result}_2 = [C_2 \times (V - V_S) + C_1 \times V_S] \times 100/L$$

At 6 h:

$$\text{Result}_3 = [C_3 \times (V - 2 \times V_S) + (C_1 + C_2) \times V_S] \times 100/L$$

At 9 h:

$$\text{Result}_4 = [C_4 \times (V - 3 \times V_S) + (C_1 + C_2 + C_3) \times V_S] \times 100/L$$

At 12 h:

$$\text{Result}_5 = [C_5 \times (V - 4 \times V_S) + (C_1 + C_2 + C_3 + C_4) \times V_S] \times 100/L$$

At 20 h:

$$\text{Result}_6 = [C_6 \times (V - 5 \times V_S) + (C_1 + C_2 + C_3 + C_4 + C_5) \times V_S] \times 100/L$$

C = as C<sub>1</sub>, C<sub>2</sub>, ..., C<sub>6</sub>, the content of niacin in the *Medium* at each time point (mg/mL)

V = volume of *Medium*, 900 mL

V<sub>S</sub> = volume of sample withdrawn at each time point (mL)

L = label claim (mg/Tablet)

**Tolerances:** The percentage of the labeled amount of niacin (C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>) dissolved at the times specified in *Table 6* and *Table 7* conforms to *Dissolution* <711>, *Acceptance Table 2*.

**Table 6. For Tablets Labeled to Contain 500 and 1000 mg/ Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 20
3	15–35
6	30–50
9	40–65
12	50–80
20	NLT 70

**Table 7. For Tablets Labeled to Contain 750 mg/ Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 16
3	15–35
6	30–50
9	40–65
12	50–75
20	NLT 75

**Test 4:** If the product complies with this test, the labeling indicates that it meets USP *Dissolution Test 4*.

**Medium:** Water; 900 mL

**Apparatus 1:** 100 rpm

**Times:** 1, 3, 6, 9, 12, and 24 h for Tablets labeled to contain 500 and 1000 mg/Tablet, and 1, 6, 12, and 24 h for Tablets labeled to contain 750 mg/Tablet; without *Medium* replacement

[NOTE—Withdraw the same volume at each time point. Pass a portion of the solution through a 0.45-μm PVDF membrane filter, discarding the first 2 mL of the filtrate.]

**Standard stock solution:** 0.5 mg/mL of USP Niacin RS in water

**Standard solution:** Dilute *Standard stock solution* with *Medium* to a final concentration of 0.02 mg/mL of USP Niacin RS.

**Sample solution**

**For Tablets labeled to contain 500 mg:** Dilute a filtered portion of the solution under test with *Medium* 25-fold.

**For Tablets labeled to contain 750 mg:** Dilute a filtered portion of the solution under test with *Medium* 40-fold.

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**For Tablets labeled to contain 1000 mg:** Dilute a filtered portion of the solution under test with *Medium* 50-fold.

**Instrumental conditions**

(See *Ultraviolet-Visible Spectroscopy* (857).)

**Mode:** UV

**Analytical wavelength:** 262 nm

**Path length:** 1 cm

**Blank:** *Medium*

**Analysis**

**Samples:** *Standard solution* and *Sample solution*  
Determine the concentration, in mg/mL, of niacin ( $C_6H_5NO_2$ ) in the sample withdrawn from the vessel at each time point:

$$\text{Result} = [(A_U - A_B)/A_S] \times C_S \times D$$

$A_U$  = absorbance of the *Sample solution*

$A_B$  = absorbance of the *Blank*

$A_S$  = absorbance of the *Standard solution*

$C_S$  = concentration of USP Niacin RS in the *Standard solution* (mg/mL)

$D$  = dilution factor for the *Sample solution*

**For Tablets labeled to contain 500 and 1000 mg:**

Calculate the percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at each time point:

At 1 h:

$$\text{Result}_1 = (C_1 \times V/L) \times 100$$

At 3 h:

$$\text{Result}_2 = [C_2 \times (V - V_S) + C_1 \times V_S] \times 100/L$$

At 6 h:

$$\text{Result}_3 = [C_3 \times (V - 2 \times V_S) + (C_1 + C_2) \times V_S] \times 100/L$$

At 9 h:

$$\text{Result}_4 = [C_4 \times (V - 3 \times V_S) + (C_1 + C_2 + C_3) \times V_S] \times 100/L$$

At 12 h:

$$\text{Result}_5 = [C_5 \times (V - 4 \times V_S) + (C_1 + C_2 + C_3 + C_4) \times V_S] \times 100/L$$

At 24 h:

$$\text{Result}_6 = [C_6 \times (V - 5 \times V_S) + (C_1 + C_2 + C_3 + C_4 + C_5) \times V_S] \times 100/L$$

**For Tablets labeled to contain 750 mg:** Calculate the percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at each time point:

At 1 h:

$$\text{Result}_1 = (C_1 \times V/L) \times 100$$

At 6 h:

$$\text{Result}_2 = [C_2 \times (V - V_S) + C_1 \times V_S] \times 100/L$$

At 12 h:

$$\text{Result}_3 = [C_3 \times (V - 2 \times V_S) + (C_1 + C_2) \times V_S] \times 100/L$$

At 24 h:

$$\text{Result}_4 = [C_4 \times (V - 3 \times V_S) + (C_1 + C_2 + C_3) \times V_S] \times 100/L$$

$C$  = as  $C_1, C_2, \dots, C_6$ , the content of niacin in the *Medium* at each time point (mg/mL)

$V$  = volume of *Medium*, 900 mL

$V_S$  = volume of sample withdrawn at each time point (mL)

$L$  = label claim (mg/Tablet)

**Tolerances:** The percentage of the labeled amount of niacin ( $C_6H_5NO_2$ ) dissolved at the times specified in *Table 8*, *Table 9*, and *Table 10* conforms to *Dissolution* (711), *Acceptance Table 2*.

**Table 8. For Tablets Labeled to Contain 500 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 15
3	17–32
6	33–48
9	48–68
12	60–80
24	NLT 80

**Table 9. For Tablets Labeled to Contain 750 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 15
6	20–40
12	48–68
24	NLT 80

**Table 10. For Tablets Labeled to Contain 1000 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 15
3	12–27
6	25–45
9	35–55
12	50–70
24	NLT 80

• **Test 5:** If the product complies with this test, the labeling indicates that it meets USP *Dissolution Test 5*.

**Medium:** Water; 900 mL

**Apparatus 1:** 100 rpm

**Times:** 1, 6, 12, and 24 h; without *Medium* replacement

[NOTE—Withdraw the same volume at each time point. Pass a portion of the solution through a 0.45- $\mu$ m nylon membrane filter, discarding the first 2 mL of the filtrate.]

**Solution A:** 1.1 mg/mL of sodium 1-heptanesulfonate in water

**Mobile phase:** Mixture of *Solution A* and methanol (70:30)

**Standard solution:** 0.84 mg/mL of USP Niacin RS in water

[NOTE—Use sonication for complete dissolution, if necessary.]

**Sample solution:** Filtered portion of the solution under test

**Chromatographic system**

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC  
**Detector:** UV 262 nm  
**Column:** 4.6-mm × 25-cm; 5-µm packing L1  
**Column temperature:** 30°  
**Flow rate:** 1.0 mL/min  
**Injection volume:** 5 µL  
**System suitability**  
**Sample:** *Standard solution*  
**Suitability requirements**  
**Tailing factor:** NMT 2.0  
**Relative standard deviation:** NMT 2.0%

**Analysis**

**Samples:** *Standard solution* and *Sample solution*  
 Determine the concentration, in mg/mL, of niacin (C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>) in the *Medium* at each time point:

$$\text{Result} = (r_U/r_S) \times C_S$$

$r_U$  = peak area of niacin from the *Sample solution*  
 $r_S$  = peak area of niacin from the *Standard solution*  
 $C_S$  = concentration of USP Niacin RS in the *Standard solution* (mg/mL)

Calculate the percentage of the labeled amount of niacin (C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>) dissolved at each time point:

At 1 h:

$$\text{Result}_1 = (C_1 \times V/L) \times 100$$

At 6 h:

$$\text{Result}_2 = [C_2 \times (V - V_S) + C_1 \times V_S] \times 100/L$$

At 12 h:

$$\text{Result}_3 = [C_3 \times (V - 2 \times V_S) + (C_1 + C_2) \times V_S] \times 100/L$$

At 24 h:

$$\text{Result}_4 = [C_4 \times (V - 3 \times V_S) + (C_1 + C_2 + C_3) \times V_S] \times 100/L$$

$C$  = as  $C_1, \dots, C_4$ , the content of niacin in the *Medium* at each time point (mg/mL)

$V$  = volume of *Medium*, 900 mL

$L$  = label claim (mg/Tablet)

$V_S$  = volume of sample withdrawn at each time point (mL)

**Tolerances:** The percentage of the labeled amount of niacin (C<sub>6</sub>H<sub>5</sub>NO<sub>2</sub>) dissolved at the times specified in *Table 11* and *Table 12* conforms to *Dissolution* <711>, *Acceptance Table 2*.

**Table 11. For Tablets Labeled to Contain 500 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 20
6	30–50
12	50–75
24	NLT 80

**Table 12. For Tablets Labeled to Contain 1000 mg/Tablet**

Time (h)	Amount Dissolved (%)
1	NMT 20
6	20–40
12	45–65
24	NLT 80

• (RB 1-Jan-2018)

• **UNIFORMITY OF DOSAGE UNITS (905):** Meet the requirements

**IMPURITIES**

• **ORGANIC IMPURITIES**

**Diluent, Mobile phase, Standard solution, Sample solution, Chromatographic system, and System suitability:** Proceed as directed in the *Assay*.

**Analysis**

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of 6-hydroxynicotinic acid or pyridine in the portion of Tablets taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak area of 6-hydroxynicotinic acid or pyridine from the *Sample solution*

$r_S$  = peak area of 6-hydroxynicotinic acid or pyridine from the *Standard solution*

$C_S$  = concentration of USP 6-Hydroxynicotinic Acid RS or pyridine in the *Standard solution* (µg/mL)

$C_U$  = nominal concentration of niacin in the *Sample solution* (µg/mL)

Calculate the percentage of any unspecified impurity in the portion of Tablets taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak area of each impurity from the *Sample solution*

$r_S$  = peak area of niacin from the *Standard solution*

$C_S$  = concentration of USP Niacin RS in the *Standard solution* (µg/mL)

$C_U$  = nominal concentration of niacin in the *Sample solution* (µg/mL)

**Acceptance criteria:** See *Table 13*.

**Table 13**

Name	Relative Retention Time	Acceptance Criteria, NMT (%)
Pyridine	0.14	0.2
6-Hydroxynicotinic acid	0.64	0.2
Niacin	1.0	—
Any unspecified impurity	—	0.1
Total impurities	—	1.0

**ADDITIONAL REQUIREMENTS**

- **PACKAGING AND STORAGE:** Preserve in tight containers.
- **LABELING:** When more than one *Dissolution* test is given, the labeling states the *Dissolution* test used only if *Test 1* is not used.
- **USP REFERENCE STANDARDS <11>**  
 USP 6-Hydroxynicotinic Acid RS  
 USP Niacin RS