

## Ethacrynic Acid Tablets

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<b>Expert Committee</b>	Small Molecules 2

In accordance with the Rules and Procedures of the Council of Experts, the Small Molecules 2 Expert Committee has revised the Ethacrynic Acid Tablets monograph. The purpose of the revision is to add *Dissolution Test 3* to accommodate FDA-approved drug products with different dissolution conditions and/or tolerances than the existing dissolution test(s).

Existing references to reagents have been updated for consistency with the reagent entry. For additional information about reagent cross references, please see the related [Compendial Notice](#).

The Ethacrynic Acid Tablets Revision Bulletin supersedes the currently official monograph.

Should you have any questions, please contact Tsion Bililign, Scientific Liaison (301-816-8286 or [tb@usp.org](mailto:tb@usp.org)).

## Ethacrynic Acid Tablets

### DEFINITION

Ethacrynic Acid Tablets contain NLT 90.0% and NMT 110.0% of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ).

### IDENTIFICATION

#### Change to read:

#### • A. ULTRAVIOLET ABSORPTION

**Diluent:** A mixture of [hydrochloric acid](#) and <sup>▲</sup> [methanol](#) (9 in 1000) (RB 1-Apr-2021)

**Standard solution:** 50 µg/mL of [USP Ethacrynic Acid RS](#) in *Diluent*

**Sample solution:** Nominally 50 µg/mL of ethacrynic acid in *Diluent* prepared as follows. Weigh a portion of finely powdered Tablets, equivalent to 50 mg of ethacrynic acid, and transfer to a separator containing 25 mL of 0.1 N [hydrochloric acid](#). Extract with two 40-mL portions of [methylene chloride](#), filter the extracts into a 100-mL volumetric flask, and dilute with [methylene chloride](#) to volume. Transfer 10.0 mL of this solution to a 100-mL volumetric flask, evaporate in a gentle current of air to dryness, and promptly dissolve the residue in a portion of *Diluent*, then dilute with *Diluent* to volume.

**Acceptance criteria:** The UV absorption spectrum of the *Sample solution* exhibits maxima and minima at the same wavelengths as those of the *Standard solution*, concomitantly measured.

#### • B.

**Sample solution:** Nominally 12.5 mg/mL of ethacrynic acid prepared as follows. Add 2 mL of 1 N [sodium hydroxide](#) to a portion of the powdered Tablets equivalent to 25 mg of ethacrynic acid.

#### Analysis

**Sample:** *Sample solution*

Heat the *Sample solution* for several minutes in a boiling water bath. Cool the solution, acidify with 0.25 mL of 18 N [sulfuric acid](#), add 0.5 mL of chromotropic acid sodium salt solution (1 in 10), then cautiously add 2 mL of [sulfuric acid TS](#).

**Acceptance criteria:** A deep violet color is produced.

### ASSAY

#### • PROCEDURE

**Solution A:** Mix 10 mL of [triethylamine](#) and about 900 mL of [water](#) in a 1-L volumetric flask. Adjust with [phosphoric acid](#) to a pH of  $6.8 \pm 0.1$ , dilute with [water](#) to volume, mix, and filter.

**Mobile phase:** [Acetonitrile](#) and *Solution A* (40:60). Filter and degas.

**Diluent:** [Acetonitrile](#) and [water](#) (40:60)

**Standard solution:** 0.5 mg/mL of [USP Ethacrynic Acid RS](#) in *Diluent*

**Sample solution:** Nominally 0.5 mg/mL of ethacrynic acid in *Diluent* prepared as follows. Transfer a portion of the powder from NLT 20 finely powdered Tablets, equivalent to about 50 mg of ethacrynic acid, to a 100-mL volumetric flask, add about 80 mL of *Diluent*, and shake or sonicate to dissolve the ethacrynic acid. Dilute with *Diluent* to volume, and mix.

#### Chromatographic system

(See [Chromatography](#) (621), [System Suitability](#).)

**Mode:** LC

**Detector:** UV 254 nm

**Column:** 3.9-mm × 30-cm; packing [L1](#)

**Flow rate:** 1 mL/min

**Injection volume:** 10 µL

### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Capacity factor,  $k'$ :** NLT 0.8

**Column efficiency:** NLT 1200 theoretical plates

**Tailing factor:** NMT 2

**Relative standard deviation:** NMT 1.0%

### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) in the portion of Tablets taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak response of ethacrynic acid from the *Sample solution*

$r_S$  = peak response of ethacrynic acid from the *Standard solution*

$C_S$  = concentration of [USP Ethacrynic Acid RS](#) in the *Standard solution* (mg/mL)

$C_U$  = nominal concentration of ethacrynic acid in the *Sample solution* (mg/mL)

**Acceptance criteria:** 90.0%–110.0%

## PERFORMANCE TESTS

### Change to read:

- [DISSOLUTION](#) <711>

#### Test 1

**Medium:** 0.1 M phosphate buffer, prepared by mixing 13.6 g of [monobasic potassium phosphate](#) and 92.2 mL of 1 N [sodium hydroxide](#) with [water](#) to obtain 1000 mL of a solution having a pH of  $8.0 \pm 0.05$ ; 900 mL

**Apparatus 2:** 50 rpm

**Time:** 45 min

**Standard solution:** A known concentration of [USP Ethacrynic Acid RS](#) in *Medium*

**Sample solution:** Pass a portion of the solution under test through a suitable filter. Dilute suitably with *Medium*.

#### Instrumental conditions

**Mode:** UV

**Analytical wavelength:** Maximum absorbance at about 277 nm

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) dissolved:

$$\text{Result} = (A_U/A_S) \times C_S \times V \times D \times (1/L) \times 100$$

$A_U$  = absorbance of the *Sample solution*

$A_S$  = absorbance of the *Standard solution*  
 $C_S$  = concentration of [USP Ethacrynic Acid RS](#) in the *Standard solution* (mg/mL)  
 $V$  = volume of *Medium*, 900 mL  
 $D$  = dilution factor for the *Sample solution*  
 $L$  = label claim (mg/Tablet)

**Tolerances:** NLT 75% (Q) of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) is dissolved.

**Test 2:** If the product complies with this test, the labeling indicates that it meets USP *Dissolution Test 2*.

**Medium:** Citrate buffer, pH 3.0 (17.2 g/L of [citric acid](#) and 5.3 g/L of [sodium citrate dihydrate](#) in [water](#), adjusted with 1 N [sodium hydroxide](#) or 1 M [acetic acid TS](#) to a pH of  $3.0 \pm 0.05$ ); 900 mL, deaerated

**Apparatus 2:** 50 rpm

**Time:** 30 min

**Solution A:** 1% [triethylamine](#) solution in [water](#), prepared as follows. Transfer a suitable aliquot of [triethylamine](#) to an appropriate volumetric flask containing 90% of the flask volume of [water](#). Adjust with [phosphoric acid](#) to a pH of  $6.8 \pm 0.1$ . Dilute with [water](#) to volume.

**Mobile phase:** [Acetonitrile](#) and *Solution A* (40:60)

**Standard stock solution:** 0.28 mg/mL of [USP Ethacrynic Acid RS](#), prepared as follows. Transfer a portion of [USP Ethacrynic Acid RS](#) to a suitable volumetric flask and add 10% of the flask volume of [methanol](#). Dilute with *Medium* to volume.

**Standard solution:** 0.028 mg/mL of [USP Ethacrynic Acid RS](#) from the *Standard stock solution*, in *Medium*

**Sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size.

### Chromatographic system

(See [Chromatography <621>](#), [System Suitability](#).)

**Mode:** LC

**Detector:** UV 277 nm

**Column:** 3.9-mm  $\times$  30-cm; 10- $\mu$ m packing [L1](#)

**Flow rate:** 1 mL/min

**Injection volume:** 10  $\mu$ L

**Run time:** NLT 2.4 times the retention time of ethacrynic acid

### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0

**Relative standard deviation:** NMT 2.0%

### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) dissolved:

$$\text{Result} = (r_U/r_S) \times C_S \times V \times (1/L) \times 100$$

$r_U$  = peak response of ethacrynic acid from the *Sample solution*

$r_S$  = peak response of ethacrynic acid from the *Standard solution*

$C_S$  = concentration of [USP Ethacrynic Acid RS](#) in the *Standard solution* (mg/mL)

$V$  = volume of *Medium*, 900 mL

$L$  = label claim (mg/Tablet)

**Tolerances:** NLT 80% (Q) of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) is dissolved.

▲ **Test 3:** If the product complies with this test, the labeling indicates that it meets USP *Dissolution Test 3*.

**Medium:** Citrate buffer, pH 3.0 (dissolve 17.4 g of citric acid, anhydrous and 2.7 g of sodium citrate dihydrate in 800 mL of water, adjust with 5 N sodium hydroxide TS to a pH of 3.0, and dilute with water to 1000 mL); 900 mL

**Apparatus 2:** 50 rpm

**Time:** 45 min

**Diluent:** Phosphate buffer, pH 8.0, prepared as follows. Dissolve 13.6 g of potassium phosphate, monobasic in 800 mL of water, add 92 mL of 1 N sodium hydroxide VS, and dilute with water to 1000 mL. Adjust with 1 N sodium hydroxide VS to a pH of 8.0.

**Standard stock solution:** 0.3 mg/mL of USP Ethacrynic Acid RS, prepared as follows. Transfer a portion of USP Ethacrynic Acid RS to a suitable volumetric flask, add *Diluent* to 60% of the flask volume, and sonicate to dissolve. Dilute with *Diluent* to volume.

**Standard solution:** ( $L/900$ ) mg/mL of USP Ethacrynic Acid RS from the *Standard stock solution* in *Medium*, where  $L$  is the label claim in mg/Tablet

**Sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size, discard a few milliliters, and collect the filtrate.

#### Instrumental conditions

(See Ultraviolet-Visible Spectroscopy (857).)

**Mode:** UV

**Analytical wavelength:** 277 nm

**Blank:** *Medium*

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) dissolved:

$$\text{Result} = (A_U/A_S) \times C_S \times V \times (1/L) \times 100$$

$A_U$  = absorbance of the *Sample solution*

$A_S$  = absorbance of the *Standard solution*

$C_S$  = concentration of USP Ethacrynic Acid RS in the *Standard solution* (mg/mL)

$V$  = volume of *Medium*, 900 mL

$L$  = label claim (mg/Tablet)

**Tolerances:** NLT 80% (Q) of the labeled amount of ethacrynic acid ( $C_{13}H_{12}Cl_2O_4$ ) is dissolved. ▲ (RB 1-

Apr-2021)

- **UNIFORMITY OF DOSAGE UNITS (905):** Meet the requirements

**Procedure for content uniformity** (if applicable)

**Diluent:** A mixture of hydrochloric acid and methanol (9 in 1000)

**Standard solution:** 50  $\mu$ g/mL of USP Ethacrynic Acid RS in *Diluent*

**Sample stock solution:** Add 1 Tablet to a 100-mL volumetric flask containing 10 mL of water, and allow to stand for 15 min, shaking occasionally until the Tablet is disintegrated. Add *Diluent* to volume and mix. Pass a portion of the mixture through a suitable filter.

**Sample solution:** Nominally 50 µg/mL of ethacrynic acid in *Diluent* from the *Sample stock solution* prepared as follows. Pipet a volume of the *Sample stock solution*, equivalent to 5 mg of ethacrynic acid, into a 100-mL volumetric flask. Dilute with *Diluent* to volume and mix.

**Instrumental conditions**

**Mode:** UV

**Analytical wavelength:** Maximum absorbance at about 269 nm

**Cell:** 1 cm

**Blank:** *Diluent*

**Analysis**

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of ethacrynic acid (C<sub>13</sub>H<sub>12</sub>Cl<sub>2</sub>O<sub>4</sub>) in the Tablet taken:

$$\text{Result} = (A_U/A_S) \times (C_S/C_U) \times 100$$

$A_U$  = absorbance of the *Sample solution*

$A_S$  = absorbance of the *Standard solution*

$C_S$  = concentration of [USP Ethacrynic Acid RS](#) in the *Standard solution* (µg/mL)

$C_U$  = nominal concentration of ethacrynic acid in the *Sample solution* (µg/mL)

**ADDITIONAL REQUIREMENTS**

- **PACKAGING AND STORAGE:** Preserve in well-closed containers.
- **LABELING:** When more than one *Dissolution* test is given, the labeling states the *Dissolution* test used only if *Test 1* is not used.
- **USP REFERENCE STANDARDS** (11).  
[USP Ethacrynic Acid RS](#)

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