

## (467) RESIDUAL SOLVENTS

### Change to read:

### ■INTRODUCTION

This general chapter applies to existing drug substances, excipients, and products. All substances and products are subject to relevant control of solvents likely to be present in a substance or product.

Where the limits to be applied comply with those given below, tests for residual solvents are not generally mentioned in specific monographs because the solvents employed may vary from one manufacturer to another.

The objective of this general chapter is to provide acceptable amounts of residual solvents in pharmaceuticals for the safety of the patient. The general chapter recommends the use of less toxic solvents and describes levels considered to be toxicologically acceptable for some residual solvents. <sup>2S (USP30)</sup>

For pharmacopeial purposes, residual solvents in pharmaceuticals are defined as organic volatile chemicals that are used or produced in the manufacture of drug substances or excipients, or in the preparation of drug products. The residual solvents are not completely removed by practical manufacturing techniques. Appropriate selection of the solvent for the synthesis of a drug substance or an excipient may enhance the yield, or determine characteristics such as crystal form, purity, and solubility. Therefore, the solvent may sometimes be a critical element in the synthetic process. This general chapter does not address solvents deliberately used as excipients nor does it address solvates. However, the content of solvents in such products should be evaluated and justified.

Because residual solvents do not provide therapeutic benefit, they should be removed, to the extent possible, to meet ingredient and product specifications, good manufacturing practices, or other quality-based requirements. Drug products should contain no higher levels of residual solvents than can be supported by safety data. <sup>2S (USP30)</sup> Solvents that are known to cause unacceptable toxicities (Class 1, *Table 1*) should be avoided in the production of drug substances, excipients, or drug products unless their use can be strongly justified in a risk-benefit assessment. <sup>2S (USP30)</sup> Solvents associated with less severe toxicity (Class 2, *Table 2*) should be limited in order to protect patients from potential adverse effects. Ideally, less toxic solvents (Class 3, *Table 3*) should be used where practical. The complete list of solvents included in this general chapter is given in *Appendix 1*. These tables and the list are not exhaustive. <sup>2S (USP30)</sup> For the purposes of this Pharmacopeia, when a manufacturer has received approval from a competent regulatory authority for the use of a new solvent not currently listed in this general chapter, it is the responsibility of that manufacturer to notify the USP regarding the identity of this solvent, the approved residual solvent limit in the article, and the appropriate test procedure for this residual solvent in the article. The USP will then address this topic in the individual monograph. When a new solvent has been approved through the ICH process, this new solvent will be added to the appropriate list in this general chapter. At that time consideration will be given for removal of the specific solvent test requirement in the individual monograph. <sup>2S (USP30)</sup>

Testing of drug substances, excipients, and drug products for residual solvents should be performed when production or purification processes are known to result in the presence of such residual solvents. It is only necessary to test for residual solvents that are used or produced in the manufacture or purification <sup>2S (USP30)</sup> of drug substances, excipients, or products. <sup>2S (USP30)</sup>

Although manufacturers may choose to test the drug product, a cumulative procedure may be used to calculate the residual solvent levels in the <sup>2S (USP30)</sup> drug product from the levels in <sup>2S (USP30)</sup> the ingredients used to produce the drug product. <sup>2S (USP30)</sup> If the calculation results in a level equal to or below that <sup>2S (USP30)</sup> pro-

vided <sup>2S (USP30)</sup> in this general chapter, no testing of the drug product for residual solvents <sup>2S (USP30)</sup> need be considered. If, however, the calculated <sup>2S (USP30)</sup> level is above the recommended level, the drug product should be tested to ascertain whether the formulation process has reduced the relevant solvent <sup>2S (USP30)</sup> level to within the acceptable amount. <sup>2S (USP30)</sup> A drug product should also be tested if a residual solvent is used during its manufacture.

<sup>2S (USP30)</sup> For the purposes of this Pharmacopeia, when a manufacturer has received approval from a competent regulatory authority for a higher level of residual solvent, it is the responsibility of that manufacturer to notify the USP regarding the identity of this solvent and the approved residual solvent limit in the article. The USP will then address this topic in the individual monograph. <sup>2S (USP30)</sup>

See *Appendix 2* for additional background information related to residual solvents.

### Change to read:

### CLASSIFICATION OF RESIDUAL SOLVENTS BY RISK ASSESSMENT

The term “tolerable daily intake” (TDI) is used by the International Program on Chemical Safety (IPCS) to describe exposure limits of toxic chemicals and the term “acceptable daily intake” (ADI) is used by the World Health Organization (WHO) and other national and international health authorities and institutes. The term “permitted daily exposure” (PDE) is defined as a pharmaceutically acceptable intake of residual solvents to avoid confusion of differing values for ADIs of the same substance.

Residual solvents <sup>2S (USP30)</sup> assessed in this general chapter are listed in *Appendix 1* by common names and structures. They were evaluated for their possible risk to human health and placed into one of three classes as follows:

Residual Solvent Class <sup>2S (USP30)</sup>	Assessment <sup>2S (USP30)</sup>
Class 1	<ul style="list-style-type: none"> <li>Solvents to be avoided <sup>2S (USP30)</sup></li> <li>Known human carcinogens</li> <li>Strongly suspected human carcinogens</li> <li>Environmental hazards</li> </ul>
Class 2	<ul style="list-style-type: none"> <li>Solvents to be limited <sup>2S (USP30)</sup></li> <li>Nongenotoxic animal carcinogens or possible causative agents of other irreversible toxicity, such as neurotoxicity or teratogenicity.</li> <li>Solvents suspected of other significant but reversible toxicities.</li> </ul>
Class 3	<ul style="list-style-type: none"> <li>Solvents with low toxic potential <sup>2S (USP30)</sup></li> <li>Solvents with low toxic potential to humans; no health-based exposure limit is needed.</li> <li>[NOTE—Class 3 residual solvents <sup>2S (USP30)</sup> have PDEs of 50 mg or more per day.]<sup>*</sup></li> </ul>

<sup>\*</sup> For residual solvents with PDEs of more than 50 mg per day, see the discussion in the section *Class 3* under *Limits of Residual Solvents*.

### Change to read:

### ■METHODS <sup>2S (USP30)</sup> FOR ESTABLISHING EXPOSURE LIMITS

The <sup>2S (USP30)</sup> method used to establish permitted daily exposures for residual solvents is presented in *Appendix 3*.

<sup>2S (USP30)</sup> For articles that are designated “for veterinary use only”, higher levels for the PDE and concentration limit may be justified in exceptional cases based upon the actual daily dose, actual target

species, and relevant toxicological data and considering consumer safety impact. For the purpose of this Pharmacopeia, when a manufacturer has received approval from a competent regulatory authority for a higher limit, it is the responsibility of that manufacturer to notify the USP regarding the approved residual solvent limit in the article and the justification. The USP will then address this topic in the individual monograph. ■<sup>2S</sup> (USP30)

**Change to read:**

**OPTIONS FOR DESCRIBING LIMITS OF CLASS 2 RESIDUAL SOLVENTS** ■<sup>2S</sup> (USP30)

Two options are available when setting limits for Class 2 residual solvents.

**Option 1**

The concentration limits in ppm stated in *Table 2* are used. They were calculated using equation (1) below by assuming a product weight of 10 g administered daily.

$$\text{Concentration (ppm)} = (1000 \mu\text{g/mL} \times \text{PDE})/\text{dose} \quad \text{■}^{\text{2S}} \text{ (USP30)}$$

Here, PDE is given in terms of mg per day, and dose is given in g per day.

These limits are considered acceptable for all drug substances, excipients, and drug products. Therefore, this option may be applied if the daily dose is not known or fixed. If all drug substances and excipients in a formulation meet the limits given in *Option 1*, these components may be used in any proportion. No further calculation is necessary provided the daily dose does not exceed 10 g. Products that are administered in doses greater than 10 g per day are to be considered under *Option 2*.

**Option 2**

It is not necessary for each component of the drug product to comply with the limits given in *Option 1*. The PDE in terms of mg per day as stated in *Table 2* can be used with the known maximum daily dose and equation (1) above to determine the concentration of residual solvent allowed in a drug product. Such limits are considered acceptable provided that it has been demonstrated that the residual solvent has been reduced to the practical minimum. The limits should be realistic in relation to analytical precision, manufacturing capability, and reasonable variation in the manufacturing process. The limits should also reflect contemporary manufacturing standards.

*Option 2* may be applied by adding the amounts of a residual solvent present in each of the components of the drug product. The sum of the amounts of solvent per day should be less than that given by the PDE.

Consider an example of the application of *Option 1* and *Option 2* to acetonitrile concentration in a drug product. The permitted daily exposure to acetonitrile is 4.1 mg per day; thus, the *Option 1* limit is 410 ppm. The maximum administered daily weight of a drug product is 5.0 g, and the drug product contains two excipients. The composition of the drug product and the calculated maximum content of residual acetonitrile are given in the following table.

Component	Amount in Formulation (g)	Acetonitrile Content (ppm)	Daily Exposure (mg)
Drug substance	0.3	800	0.24
Excipient 1	0.9	400	0.36
Excipient 2	3.8	800	3.04
Drug product	5.0	728	3.64

Excipient 1 meets the *Option 1* limit, but the drug substance, excipient 2, and drug product do not meet the *Option 1* limit. Nevertheless, the drug product meets the *Option 2* limit of 4.1 mg per day and thus conforms to the acceptance criteria in this General Chapter.

Consider another example using acetonitrile as the residual solvent. The maximum administered daily weight of a drug product is 5.0 g, and the drug product contains two excipients. The composition of the drug product and the calculated maximum content of residual acetonitrile are given in the following table.

Component	Amount in Formulation (g)	Acetonitrile Content (ppm)	Daily Exposure (mg)
Drug substance	0.3	800	0.24
Excipient 1	0.9	2000	1.80
Excipient 2	3.8	800	3.04
Drug product	5.0	1016	5.08

In this example, the drug product meets neither the *Option 1* nor the *Option 2* limit according to this summation. ■<sup>2S</sup> (USP30) The manufacturer could test the drug product to determine if the formulation process reduced the level of acetonitrile. If the level of acetonitrile was not reduced to the allowed limit during formulation, the product fails to meet the solvent limits as described in this chapter and the manufacturer of the drug product should take other steps to reduce the amount of acetonitrile in the drug product. In some instances the manufacturer may have received approval from a competent regulatory authority for such a higher level of residual solvent. If this is the case, it is the responsibility of that manufacturer to notify the USP regarding the identity of this solvent and the approved residual solvent limit in the article. The USP will then address this topic in the individual monograph. ■<sup>2S</sup> (USP30)

**Add the following:**

**ANALYTICAL PROCEDURES**

Residual solvents are typically determined using chromatographic techniques such as gas chromatography. Compendial methods for testing for residual solvent content are described under the *Identification, Control, and Quantification of Residual Solvents* section of this general chapter. The *General Notices* discuss the use of other methods in special circumstances (see *Procedures in Tests and Assays*). If Class 3 solvents are present, a nonspecific method such as loss on drying may be used. ■<sup>2S</sup> (USP30)

**Add the following:**

**REPORTING LEVELS OF RESIDUAL SOLVENTS**

Manufacturers of pharmaceutical products need certain information about the content of residual solvents in drug substances or excipients in order to meet the criteria of this general chapter. The following statements are given as acceptable examples of the information that could be provided from a supplier of drug substances or excipients to a pharmaceutical manufacturer. The supplier might choose one of the following as appropriate:

- Only Class 3 solvents are likely to be present. Loss on drying is less than 0.5%.
- Only Class 2 solvents X, Y, ... are likely to be present. All are below the *Option 1* limit. (Here the supplier would name the Class 2 solvents represented by X, Y, ...)
- Only Class 2 solvents X, Y, ... and Class 3 solvents are likely to be present. Residual Class 2 solvents are below the *Option 1* limit and residual Class 3 solvents are below 0.5%.

The phrase “likely to be present” as used in the above examples refers to the solvent used or produced in the final manufacturing step and to solvents that are used or produced in earlier manufacturing steps and not removed consistently by a validated process.

If Class 1 solvents are likely to be present, they should be identified and quantified. If solvents of Class 2 or 3 are present at greater than their *Option 1* limits or 0.5%, respectively, they should be identified and quantified. <sup>■2S (USP30)</sup>

### Change to read:

## LIMITS OF RESIDUAL SOLVENTS

### Ethylene Oxide

[NOTE—The test for ethylene oxide is conducted only where specified in the individual monograph.] The standard solution parameters and the procedure for determination are described in the individual monograph. Unless otherwise specified in the individual monograph, the limit is 10 µg per g.

### Class 1 <sup>■(solvents to be avoided)</sup> <sup>■2S (USP30)</sup>

Class 1 residual solvents (*Table 1*) should not be employed in the manufacture of drug substances, excipients, and drug products because of the unacceptable toxicities or deleterious environmental effects of these residual solvents. However, if their use in order to produce a medicinal product <sup>■</sup>with a significant therapeutic advance <sup>■2S (USP30)</sup> is unavoidable, their levels should be restricted as shown in *Table 1*, unless otherwise stated in the individual monograph. The solvent 1,1,1-trichloroethane is included in *Table 1* because it is an environmental hazard. The stated limit of 1500 ppm is based on <sup>■</sup>a review of <sup>■2S (USP30)</sup> safety data.

When Class 1 residual solvents are used or produced in the manufacture or purification of a drug substance, excipient, or drug product <sup>■</sup>and are not removed by the process, <sup>■2S (USP30)</sup> these solvents should be identified and quantified. The procedures described in the *Identification, Control, and Quantification of Residual Solvents* section of this general chapter are to be applied wherever possible. Otherwise an appropriate validated procedure is to be employed. <sup>■2S (USP30)</sup>

**Table 1. Class 1 Residual Solvents**  
<sup>■(solvents that should be avoided)</sup> <sup>■2S (USP30)</sup>

Solvent	Concentration Limit (ppm)	Concern
Benzene	2	Carcinogen
Carbon tetrachloride	4	Toxic and environmental hazard
1,2-Dichloroethane	5	Toxic
1,1-Dichloroethene	8	Toxic
1,1,1-Trichloroethane	1500	Environmental hazard

### Class 2

Class 2 residual solvents (*Table 2*) should be limited in drug substances, excipients, and drug products because of the inherent toxicities of the residual solvents. PDEs are given to the nearest 0.1 mg per day, and concentrations are given to the nearest 10 ppm. The stated values do not reflect the necessary analytical precision of the determination procedure. Precision should be determined as part of the procedure validation.

If Class 2 residual solvents are present at greater than their *Option 1* limits, they should be identified and quantified. The procedures described in the *Identification, Control, and Quantification of*

*Residual Solvents* section of this general chapter are to be applied wherever possible. Otherwise an appropriate validated procedure is to be employed. <sup>■2S (USP30)</sup>

[NOTE—The following Class 2 residual solvents are not readily detected by the headspace injection conditions described in the *Identification, Control, and Quantification of Residual Solvents* section of this general chapter: formamide, 2-ethoxyethanol, 2-methoxyethanol, ethylene glycol, *N*-methylpyrrolidone, and sulfolane. Other appropriate validated procedures are to be employed for the quantification of these residual solvents. Such procedures shall be submitted to the USP for review and possible inclusion in the relevant individual monograph. <sup>■</sup>In addition, USP Residual Solvent Class 2—Mixture C RS can be used to develop an alternative procedure. <sup>■2S (USP30)</sup>]

**Table 2. Class 2 Residual Solvents**

Solvent	PDE (mg/day)	Concentration Limit (ppm)
Acetonitrile	4.1	410
Chlorobenzene	3.6	360
Chloroform	0.6	60
Cyclohexane	38.8	3880
1,2-Dichloroethene	18.7	1870
1,2-Dimethoxyethane	1.0	100
<i>N,N</i> -Dimethylacetamide	10.9	1090
<i>N,N</i> -Dimethylformamide	8.8	880
1,4-Dioxane	3.8	380
2-Ethoxyethanol	1.6	160
Ethylene glycol	6.2	620
Formamide	2.2	220
Hexane	2.9	290
Methanol	30.0	3000
2-Methoxyethanol	0.5	50
Methylbutylketone	0.5	50
Methylcyclohexane	11.8	1180
Methylene chloride	6.0	600
<i>N</i> -Methylpyrrolidone	5.3	530
Nitromethane	0.5	50
Pyridine	2.0	200
Sulfolane	1.6	160
Tetrahydrofuran	7.2	720
Tetralin	1.0	100
Toluene	8.9	890
Trichloroethylene	0.8	80
Xylene*	21.7	2170

\* Usually 60% *m*-xylene, 14% *p*-xylene, 9% *o*-xylene with 17% ethyl benzene

### Class 3

Class 3 residual solvents (*Table 3*) may be regarded as less toxic and of lower risk to human health than Class 1 and Class 2 residual solvents. Class 3 includes no solvent known as a human health hazard at levels normally accepted in pharmaceuticals. However, there are no long-term toxicity or carcinogenicity studies for many of the residual solvents in Class 3. Available data indicate that they are less toxic in acute or short-term studies and negative in genotoxicity studies.

<sup>■</sup>It is considered that amounts of these residual solvents of 50 mg per day or less (corresponding to 5000 ppm or 0.5% under *Option 1*) would be acceptable without justification. Higher amounts may also be acceptable provided they are realistic in relation to manufacturing capability and good manufacturing practice. For the purposes of this Pharmacopeia, when a manufacturer has received approval from a competent regulatory authority for such a higher level of residual solvent, it is the responsibility of that manufacturer to notify the USP regarding the identity of this solvent and the approved residual solvent limit in the article. The USP will then address this topic in the individual monograph. <sup>■2S (USP30)</sup> If a Class 3 solvent limit in an individual monograph is greater than 50 mg per day, that residual solvent should be identified and quantified. The procedures described in the *Identification, Control, and Quantification of*

*Residual Solvents* section of this General Chapter, with appropriate modifications to the standard solutions, are to be applied wherever possible. Otherwise an appropriate validated procedure is to be employed. ■<sub>2S</sub> (USP30)

**Table 3. Class 3 Residual Solvents**  
(limited by GMP or other quality-based requirements in drug substances, excipients, and drug products)

Acetic acid	Heptane
Acetone	Isobutyl acetate
Anisole	Isopropyl acetate
1-Butanol	Methyl acetate
2-Butanol	3-Methyl-1-butanol
Butyl acetate	Methylethylketone
<i>tert</i> -Butylmethyl ether	Methylisobutylketone
Cumene	2-Methyl-1-propanol
Dimethyl sulfoxide	Pentane
Ethanol	1-Pentanol
Ethyl acetate	1-Propanol
Ethyl ether	2-Propanol
Ethyl formate	Propyl acetate
Formic acid	

### Other Residual Solvents

The residual solvents listed in *Table 4* may also be of interest to manufacturers of drug substances, excipients, or drug products. However, no adequate toxicological data on which to base a PDE was found.

**Table 4. Other Residual Solvents**  
(for which no adequate toxicological data was found)

1,1-Diethoxypropane	Methyl isopropyl ketone
1,1-Dimethoxymethane	Methyltetrahydrofuran
2,2-Dimethoxypropane	Solvent hexane
Isooctane	Trichloroacetic acid
Isopropyl ether	Trifluoroacetic acid

### Change to read:

## IDENTIFICATION, CONTROL, AND QUANTIFICATION OF RESIDUAL SOLVENTS

•Whenever possible, the substance under test needs to be dissolved to release the residual solvent. Because the USP deals with drug products, as well as active ingredients and excipients, it may be acceptable that in cases some of the components of the formulation will not dissolve completely. In those cases, the drug product may first need to be pulverized into a fine powder so that any residual solvent that may be present can be released. This operation should be as fast as possible to prevent the loss of volatile solvents during the procedure.●<sub>3</sub>

NOTE—The organic-free water specified in the following procedures produces no significantly interfering peaks when chromatographed.

### Class 1 and Class 2 Residual Solvents

•The following procedures are useful to identify and quantify residual solvents when the information regarding which solvents are likely to be present in the material is not available. When the

information about the presence of specific residual solvents is available, only *Procedure C* is needed to quantify the amount of residual solvents present.●<sub>3</sub>

### WATER-SOLUBLE ARTICLES

#### Procedure A—

*Class 1 Standard Stock Solution*—Transfer 1.0 mL of USP Class 1 Residual Solvents Mixture RS to a 100-mL volumetric flask, add 9 mL of dimethyl sulfoxide, dilute with water to volume, and mix. Transfer 1.0 mL of this solution to a 100-mL volumetric flask, dilute with water to volume, and mix. Transfer 1.0 mL of this solution to a 10-mL volumetric flask, dilute with water to volume, and mix.

*Class 1 Standard Solution*—Transfer 1.0 mL of *Class 1 Standard Stock Solution* to an appropriate headspace vial, add 5.0 mL of water, apply the stopper, cap, and mix.

*Class 2 Standard Stock Solutions*—Transfer 1.0 mL of USP Residual Solvents Class 2—Mixture A RS to a 100-mL volumetric flask, dilute with water to volume, and mix. This is *Class 2 Standard Stock Solution A*. Transfer 1.0 mL of USP Residual Solvents Class 2—Mixture B RS to a 100-mL volumetric flask, dilute with water to volume, and mix. This is *Class 2 Standard Stock Solution B*.

*Class 2 Mixture A Standard Solution*—Transfer 1.0 mL of *Class 2 Standard Stock Solution A* to an appropriate headspace vial, add 5.0 mL of water, apply the stopper, cap, and mix.

*Class 2 Mixture B Standard Solution*—Transfer 5.0 mL of *Class 2 Standard Stock Solution B* to an appropriate headspace vial, add 1.0 mL of water, apply the stopper, cap, and mix.

*Test Stock Solution*—Transfer about 250 mg of the article under test, accurately weighed, to a 25-mL volumetric flask, dissolve in and dilute with water to volume, and mix.

*Test Solution*—Transfer 5.0 mL of *Test Stock Solution* to an appropriate headspace vial, add 1.0 mL of water, apply the stopper, cap, and mix.

*Class 1 System Suitability Solution*—Transfer 1.0 mL of *Class 1 Standard Stock Solution* to an appropriate headspace vial, add 5.0 mL of *Test Stock Solution*, apply the stopper, cap, and mix.

*Chromatographic System* (see *Chromatography* (621))—The gas chromatograph is equipped with a flame-ionization detector, a 0.32-mm × 30-m fused-silica column coated with a 1.8-μm layer of phase G43 or a 0.53-mm × 30-m wide-bore column coated with a 3.0-μm layer of phase G43. The carrier gas is nitrogen or helium with a linear velocity of about 35 cm per second, and a split ratio of 1:5. •[NOTE—Split ratio can be modified in order to optimize sensitivity.]●<sub>3</sub> The column temperature is maintained at 40° for 20 minutes, then raised at a rate of 10° per minute to 240°, and maintained at 240° for 20 minutes. The injection port and detector temperatures are maintained at 140° and 250°, respectively. Chromatograph the *Class 1 Standard Solution*, *Class 1 System Suitability Solution*, and *Class 2 Mixture A Standard Solution*, and record the peak responses as directed for *Procedure*: the signal-to-noise ratio of 1,1,1-trichloroethane in the *Class 1 Standard Solution* is not less than 5; the signal-to-noise ratio of each peak in the *Class 1 System Suitability Solution* is not less than 3; and the resolution, *R*, between acetonitrile and methylene chloride in the *Class 2 Mixture A Standard Solution* is not less than 1.0.

*Procedure*—Separately inject (following one of the headspace operating parameter sets described in the table below) equal volumes of headspace (about 1.0 mL) of the *Class 1 Standard Solution*, *Class 2 Mixture A Standard Solution*, *Class 2 Mixture B Standard Solution*, and the *Test Solution* into the chromatograph, record the chromatograms, and measure the responses for the major peaks. •If a peak response of any peak, other than a peak for 1,1,1-trichloroethane, in the *Test Solution* is greater than or equal to a corresponding peak in either the *Class 1 Standard Solution* or either of the two *Class 2 Mixture Standard Solutions*, or a peak response of 1,1,1-trichloroethane is greater than or equal to 150 times the peak response corresponding to 1,1,1-trichloroethane in the *Class*

1 Standard Solution, proceed to Procedure B to verify the identity of the peak; otherwise the article meets the requirements of this test.●●

**Table 5. Headspace Operating Parameters**

	Headspace Operating Parameter Sets		
	1	2	3
Equilibration temperature (°)	80	105	80
Equilibration time (min.)	60	45	45
Transfer-line temperature (°)	85	110	105
Carrier gas: nitrogen or helium at an appropriate pressure			
Pressurization time (s)	30	30	30
Injection volume (mL)	1	1	1

#### Procedure B—

Class 1 Standard Stock Solution, Class 1 Standard Solution, Class 2 Standard Stock Solutions, Class 2 Mixture A Standard Solution, Class 2 Mixture B Standard Solution, Test Stock Solution, Test Solution, and Class 1 System Suitability Solution—Prepare as directed for Procedure A.

●●  
**Chromatographic System** (see *Chromatography* (621))—The gas chromatograph is equipped with a flame-ionization detector, a 0.32-mm × 30-m fused-silica column coated with a 0.25-μm layer of phase G16, or a 0.53-mm × 30-m wide-bore column coated with a 0.25-μm layer of phase G16. The carrier gas is nitrogen or helium with a linear velocity of about 35 cm per second and a split ratio of 1:5. ●[NOTE—Split ratio can be modified in order to optimize sensitivity.]● The column temperature is maintained at 50° for 20 minutes, then raised at a rate of 6° per minute to 165°, and maintained at 165° for 20 minutes. The injection port and detector temperatures are maintained at 140° and 250°, respectively. Chromatograph the Class 1 Standard Solution and the Class 1 System Suitability Solution, ●●, and record the peak responses as directed for Procedure: the signal-to-noise ratio of benzene in the Class 1 Standard Solution is not less than 5; the signal-to-noise ratio of each peak in the Class 1 System Suitability Solution is not less than 3; and the resolution, *R*, between acetonitrile and ●cis-dichloroethene in the Class 2 Mixture A Standard Solution●, is not less than 1.0.

**Procedure**—Separately inject (following one of the headspace operating parameter sets described in Table 5) equal volumes of headspace (about 1.0 mL) of the Class 1 Standard Solution, the Class 2 Mixture A Standard Solution, the Class 2 Mixture B Standard Solution, and the Test Solution into the chromatograph, record the chromatograms, and measure the responses for the major peaks. If the peak response(s) in the Test Solution of the peak(s) identified in Procedure A is/are greater than or equal to a corresponding peak(s) in either the Class 1 Standard Solution or either of the two Class 2 Mixture Standard Solutions, proceed to Procedure C to quantify the peak(s); otherwise the article meets the requirements of this test.

#### Procedure C—

Class 1 Standard Stock Solution, Class 1 Standard Solution, Class 2 Standard Stock Solution A, Class 2 Mixture A Standard Solution, Test Stock Solution, Test Solution, and Class 1 System Suitability Solution—Prepare as directed for Procedure A.

**Standard Solution**—[NOTE—●Prepare a separate Standard Solution for each peak identified and verified by Procedures A and B. For the Class 1 solvents other than 1,1,1-trichloroethane, prepare the first dilution as directed for the first dilution under Class 1 Standard Stock Solution in Procedure A.●] Transfer an accurately measured volume of each individual USP Reference Standard corresponding to each residual solvent peak identified and verified by Procedures A and B to a suitable container, and dilute quantitatively, and stepwise if necessary, with water to obtain a solution having a final concentration of 1/20 of the value stated in Table 1 or 2 (under *Concentration Limit*). Transfer 1.0 mL of this solution to an appropriate headspace vial, add 5.0 mL of water, apply the stopper, cap, and mix.

**Spiked Test Solution**—[NOTE—Prepare a separate Spiked Test Solution for each peak identified and verified by Procedures A and B.] Transfer 5.0 mL of Test Stock Solution to an appropriate headspace vial, add 1.0 mL of the Standard Solution, apply the stopper, cap, and mix.

**Chromatographic System** (see *Chromatography* (621))—[NOTE—If the results of the chromatography from Procedure A are found to be inferior to those found with Procedure B, the Chromatographic System from Procedure B may be substituted.] The gas chromatograph is equipped with a flame-ionization detector, a 0.32-mm × 30-m fused-silica column coated with a 1.8-μm layer of phase G43 or a 0.53-mm × 30-m wide-bore column coated with a 3.0-μm layer of phase G43. The carrier gas is nitrogen or helium with a linear velocity of about 35 cm per second, and a split ratio of 1:5. ●[NOTE—The split ratio can be modified in order to optimize sensitivity.]● The column temperature is maintained at 40° for 20 minutes, then raised at a rate of 10° per minute to 240°, and maintained at 240° for 20 minutes. The injection port and detector temperatures are maintained at 140° and 250°, respectively. Chromatograph the Class 1 Standard Solution, the Class 1 System Suitability Solution, and the Class 2 Mixture A Standard Solution, and record the peak responses as directed for Procedure: the signal-to-noise ratio of 1,1,1-trichloroethane in the Class 1 Standard Solution is not less than 5; the signal-to-noise ratio of each peak in the Class 1 System Suitability Solution is not less than 3; and the resolution, *R*, between acetonitrile and methylene chloride in the Class 2 Mixture A Standard Solution is not less than 1.0.

**Procedure**—Separately inject (following one of the headspace operating parameters described in Table 5) equal volumes of headspace (about 1.0 mL) of the Standard Solution, the Test Solution, and the Spiked Test Solution into the chromatograph, record the chromatograms, and measure the responses for the major peaks. Calculate the amount, in ppm, of each residual solvent found in the article under test by the formula:

$$5(C/W)[r_U/(r_{ST} - r_U)]$$

in which *C* is the concentration, in ●μg per mL●, of the appropriate USP Reference Standard in the Standard Solution; *W* is the weight, in g, of the article under test taken to prepare the Test Stock Solution; and *r<sub>U</sub>* and *r<sub>ST</sub>* are the peak responses of each residual solvent obtained from the Test Solution and the Spiked Test Solution, respectively.

#### WATER-INSOLUBLE ARTICLES

●**Procedure A**—[NOTE—Dimethyl sulfoxide may be substituted as an alternative solvent to dimethylformamide.]

**Class 1 Standard Stock Solution**—Transfer 1.0 mL of USP Class 1 Residual Solvents Mixture RS to a 100-mL volumetric flask previously filled with about 80 mL of dimethylformamide, dilute with dimethylformamide to volume, and mix. Transfer 1.0 mL of this solution to a 100-mL volumetric flask, previously filled with about 80 mL of dimethylformamide, dilute with dimethylformamide to volume, and mix (reserve a portion of this solution for the Class 1 System Suitability Solution). Transfer 1.0 mL of this solution to a 10-mL volumetric flask, dilute with dimethylformamide to volume, and mix.

**Class 1 Standard Solution**—Transfer 1.0 mL of Class 1 Standard Stock Solution to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Class 2 Standard Stock Solutions**—Transfer 1.0 mL of USP Residual Solvents Class 2—Mixture A RS to a 100-mL volumetric flask, previously filled with about 80 mL of dimethylformamide, dilute with dimethylformamide to volume, and mix. This is Class 2 Standard Stock Solution A. Transfer 0.5 mL of USP Residual Solvents Class 2—Mixture B RS to a 10-mL volumetric flask, dilute with dimethylformamide to volume, and mix. This is Class 2 Standard Stock Solution B.

**Class 2 Mixture A Standard Solution**—Transfer 1.0 mL of Class 2 Standard Stock Solution A to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Class 2 Mixture B Standard Solution**—Transfer 1.0 mL of *Class 2 Standard Stock Solution B* to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Test Stock Solution**—Transfer about 500 mg of the article under test, accurately weighed, to a 10-mL volumetric flask, dissolve in and dilute with dimethylformamide to volume, and mix.

**Test Solution**—Transfer 1.0 mL of *Test Stock Solution* to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Class 1 System Suitability Solution**—Mix 5 mL of *Test stock solution* with 0.5 mL of the intermediate dilution reserved from *Class 1 Standard Stock Solution*. Transfer 1.0 mL of this solution to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Chromatographic System** (see *Chromatography* (621))—The gas chromatograph is equipped with a flame-ionization detector, a 0.53-mm × 30-m wide-bore column coated with a 3.0-μm layer of phase G43. The carrier gas is helium with a linear velocity of about 35 cm per second, and a split ratio of 1:3 [NOTE—Split ratio can be modified in order to optimize sensitivity.] The column temperature is maintained at 40° for 20 minutes, then raised at a rate of 10° per minute to 240°, and maintained at 240° for 20 minutes. The injection port and detector temperatures are maintained at 140° and 250°, respectively. Chromatograph the *Class 1 Standard Solution*, *Class 1 System Suitability Solution*, and *Class 2 Mixture A Standard Solution*, and record the peak responses as directed for *Procedure*: the signal-to-noise ratio of 1,1,1-trichloroethane in the *Class 1 Standard Solution* is not less than 5; the signal-to-noise ratio of each peak in the *Class 1 System Suitability Solution* is not less than 3; and the resolution, *R*, between acetonitrile and methylene chloride in the *Class 2 Mixture A Standard Solution* is not less than 1.0.

**Procedure**—Separately inject (use headspace operating parameters 3 in Table 5 with a vial pressure of 10 psi) equal volumes of headspace (about 1.0 mL) of the *Class 1 Standard Solution*, *Class 2 Mixture A Standard Solution*, *Class 2 Mixture B Standard Solution*, and *Test Solution* into the chromatograph, record the chromatograms, and measure the responses for the major peaks. If a peak response of any peak, other than a peak for 1,1,1-trichloroethane, in the *Test Solution* is greater than or equal to a corresponding peak in either the *Class 1 Standard Solution* or either of the two *Class 2 Mixture Standard Solutions*, or a peak response of 1,1,1-trichloroethane is greater than or equal to 150 times the peak response corresponding to 1,1,1-trichloroethane in the *Class 1 Standard Solution*, proceed to *Procedure B* to verify the identity of the peak; otherwise the article meets the requirements of this test.

#### Procedure B—

*Class 1 Standard Stock Solution*, *Class 1 Standard Solution*, *Class 1 System Suitability Solution*, *Class 2 Standard Stock Solutions*, *Class 2 Mixture A Standard Solution*, and *Class 2 Mixture B Standard Solution*, *Test Stock Solution*, and *Test Solution*—Proceed as directed for *Procedure A*.

**Chromatographic System**—Proceed as directed for *Procedure B* under *Water-Soluble Articles* with a split ratio of 1:3 [NOTE—The split ratio can be modified in order to optimize sensitivity.]

**Procedure**—Separately inject (use headspace operating parameters 3 in Table 5 with a vial pressure of 10 psi) equal volumes of headspace (about 1.0 mL) of the *Class 1 Standard Solution*, *Class 2 Mixture A Standard Solution*, *Class 2 Mixture B Standard Solution*, and *Test Solution*, into the chromatograph, record the chromatograms, and measure the responses for the major peaks. If the peak response(s) in *Test Solution* of the peak(s) identified in *Procedure A* is/are greater than or equal to a corresponding peak(s) in either the *Class 1 Standard Solution* or any of the two *Class 2 Mixture Standard Solutions*, proceed to *Procedure C* to quantify the peak(s); otherwise the article meets the requirements of this test.

#### Procedure C—

*Class 1 Standard Stock Solution*, *Class 1 Standard Solution*, *Class 1 System Suitability Solution*, *Class 2 Standard Stock Solution A*, and *Class 2 Mixture A Standard Solution*—Proceed as directed for *Procedure A*.

**Standard Stock Solution**—[NOTE—Prepare a separate Standard Solution for each peak identified and verified by *Procedures A* and *B*.] Transfer an accurately measured volume of each individual USP

Reference Standard corresponding to each residual solvent peak identified and verified by *Procedures A* and *B* to a suitable container, and dilute quantitatively, and stepwise if necessary, with water to obtain a solution having a final concentration of 1/20 of the value stated in Table 1 or Table 2 (under *Concentration Limit*).

**Standard Solution**—Transfer 1.0 mL of the *Standard Stock Solution* to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Test Stock Solution**—Proceed as directed for *Procedure A*.

**Test Solution**—Transfer 1.0 mL of the *Test Stock Solution* to an appropriate headspace vial, containing 5.0 mL of water, apply the stopper, cap, and mix.

**Spiked Test Solution**—[NOTE—Prepare a separate *Spiked Test Solution* for each peak identified and verified by *Procedures A* and *B*.] Transfer 1.0 mL of *Test Stock Solution* to an appropriate headspace vial, add 1 mL of *Standard Stock Solution* and 4.0 mL of water, apply the stopper, cap, and mix.

**Chromatographic System**—Proceed as directed for *Procedure C* under *Water-Soluble Articles*.

**Procedure**—Separately inject (use headspace operating parameters 3 in Table 5 with a vial pressure of 10 psi) equal volumes of headspace (about 1.0 mL) of the *Standard Solution*, *Test Solution*, and *Spiked Test Solution* into the chromatograph, record the chromatograms, and measure the responses for the major peaks. Calculate the amount, in ppm, of each residual solvent found in the article under test by the formula:

$$10(C/W)[r_U/(r_{ST} - r_U)]$$

in which *C* is the concentration, in μg per mL, of the appropriate USP Reference Standard in the *Standard Solution*; *W* is the weight, in g, of the article under test taken to prepare the *Test Stock Solution*; and *r<sub>U</sub>* and *r<sub>ST</sub>* are the peak responses of each residual solvent obtained from *Test Solution* and *Spiked Test Solution*, respectively.●●

## Class 3 Residual Solvents

●If Class 3 solvents are present, the level of residual solvents may be determined as directed under *Loss on Drying* (731) when the monograph for the article under test contains a loss on drying procedure, or a specific determination of the solvent may be made. If there is no loss on drying procedure in the monograph for the article under test or if ●, a Class 3 solvent limit in an individual monograph is greater than 50 mg per day (corresponding to 5000 ppm or 0.5% under *Option 1*), ●the individual Class 3 residual solvent or solvents present in the article under test●, should be identified and quantified, and the procedures as described above, with appropriate modifications to the standard solutions, are to be applied wherever possible. Otherwise an appropriate validated procedure is to be employed. ●● USP Reference Standards, where available, should be used in these procedures. A flow diagram for the application of residual solvent limit tests is shown in *Figure 1*.

### Change to read:

## GLOSSARY

■ **Acceptable daily intake (ADI)**: The maximum acceptable intake of toxic chemicals per day. This term is used by the World Health Organization (WHO).●2S (USP30)

● **Genotoxic carcinogens**: Carcinogens that produce cancer by affecting genes or chromosomes.

● **Lowest-observed-effect level (LOEL)**: The lowest dose of a substance in a study or group of studies that produces biologically significant increases in frequency or severity of any effects in exposed humans or animals.

● **Modifying factor**: A factor determined by professional judgment of a toxicologist and applied to bioassay data so that the data can be safely related to humans.

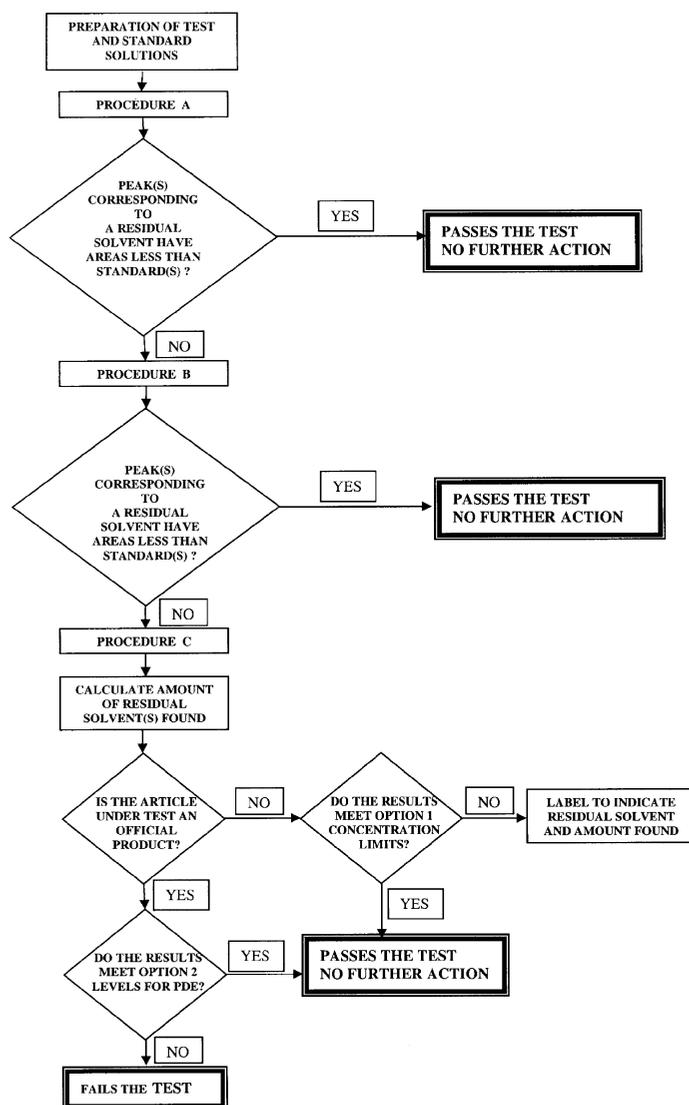


Figure 1. Diagram relating to the identification of residual solvents and the application of limit tests.

**Neurotoxicity:** The ability of a substance to cause adverse effects on the nervous system.

**No-observed-effect level (NOEL):** The highest dose of a substance at which there are no biologically significant increases in frequency or severity of any effects in exposed humans or animals.

**Permitted daily exposure (PDE):** The maximum acceptable intake per day of a residual solvent in pharmaceutical products.

**Reversible toxicity:** The occurrence of harmful effects that are caused by a substance and that disappear after exposure to the substance ends.

**Strongly suspected human carcinogen:** A substance for which there is no epidemiological evidence of carcinogenesis but for which there are positive genotoxicity data and clear evidence of carcinogenesis in rodents.

**Teratogenicity:** The occurrence of structural malformations in a developing fetus when a substance is administered during pregnancy.

■ **Tolerable daily intake (TDI):** Tolerable daily exposure to toxic chemicals. Term used by the International Program on Chemical Safety (IPCS). ■<sub>2S</sub> (USP30)